

UNIT I

PHARMACOKINETICS

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ABSORPTION, DISTRIBUTION, METABOLISM AND EXCRETION

- × relevant to ALL drugs
- × large research/development area
- × frequent cause of failure of treatment
 - + failure of compliance
 - + failure to achieve effective level
 - + produce toxic effects
 - + drug interactions (*****)
- × can enhance patient satisfaction with treatment
- × understand different dosage forms available

OVERVIEW - ADME

Most drugs :

- enter the body (by mouth or injection or...) - must cross barriers to entry (skin, gut wall, alviolar membrane.....)
- are distributed by the blood to the site of action - intra- or extra-cellular - cross barriers to distribution (capillaries, cell wall....) - distribution affects concentration at site of action and sites of excretion and biotransformation
- are biotransformed perhaps to several different compounds by enzymes evolved to cope with natural materials - this may increase, decrease or change drug actions
- are excreted (by kidney or) which removes them and/or their metabolites from the body

Pharmacokinetics is the quantification of these processes

OVERVIEW - ABSORPTION

Some drugs work outside the body (barrier creams, some laxatives)
but most must:

- ✗ enter the body:

Given by: ENTERAL - oral, sublingual, buccal, rectal

PARENTERAL sc, im, iv, it

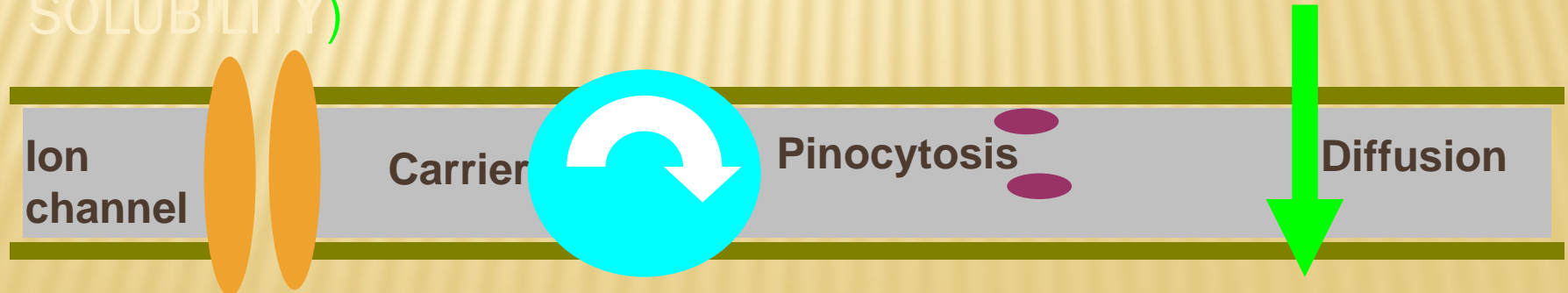
- ✗ cross lipid barriers / cell walls:

gut wall, capillary wall, cell wall, blood brain barrier

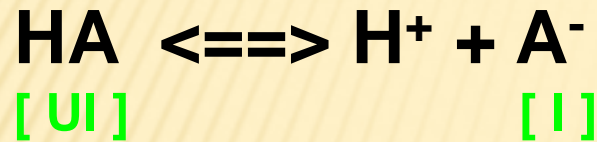
--- get into the body and (after distribution) to reach
the cellular target ---

PASSAGE THROUGH LIPID MEMBRANES

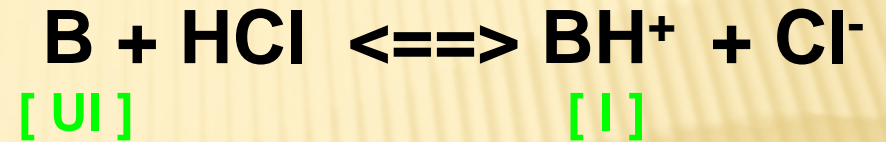
- ✗ diffusion through gaps between cells (glomerulus = 68K; capillary 30K; NB brain capillary - tight junction)
- ✗ passage through the cell membrane
 - + diffuse through pore (very small; use dependent)
 - + carrier mediated transport (specific, saturable; Fe in gut; L-DOPA at blood-brain barrier)
 - + pinocytosis (insulin in CNS; botulinum toxin in gut)
 - + diffusion through lipid of cell membrane (depends on AREA, DIFFUSION GRADIENT, DIFFUSION COEFFICIENT, LIPID SOLUBILITY)



LIPID SOLUBILITY :WEAK ACIDS AND WEAK BASES



$$\text{pKa} = \text{pH} + \log(\text{HA}/\text{A}^-)$$



$$\text{pKa} = \text{pH} + \log(\text{BH}^+/\text{B})$$

ASPIRIN pKa = 4.5 (weak acid)
100mg orally

STRYCHNINE pKa = 9.5 (weak base)
100mg orally

$$0.1 = [\text{I}]$$

Stomach
pH = 2



Blood
pH = 7.4

$$99.9 = [\text{UI}] \rightarrow [\text{UI}]$$



**Aspirin is reasonably absorbed
from stomach (fast action)**

$$99.9 = [\text{I}]$$

Stomach
pH = 2



Blood
pH = 7.4

$$0.1 = [\text{UI}] \rightarrow [\text{UI}]$$



**Strychnine not absorbed until
enters duodenum**

ROUTES OF ADMINISTRATION

- ✗ Enteral; oral, sub-lingual (buccal), rectal. Note soluble, enteric coated or slow release formulations
- ✗ Parenteral; iv, im, sc, id, it, etc. Different rates of absorption, different plasma peaks. Note iv infusors
- ✗ Skin; for local or systemic effect - note patches
- ✗ Lungs; inhalation; local or systemic effect?
- ✗ Vaginal; (usually local)
- ✗ Eye; (usually local)

FACTORS AFFECTING ORAL ABSORPTION

- Disintegration of dosage form
- Dissolution of particles
- Chemical stability of drug
- Stability of drug to enzymes
- Motility and mixing in GI tract
- Presence and type of food
- Passage across GI tract wall
- Blood flow to GI tract
- Gastric emptying time
- FORMULATION

BIOAVAILABILITY

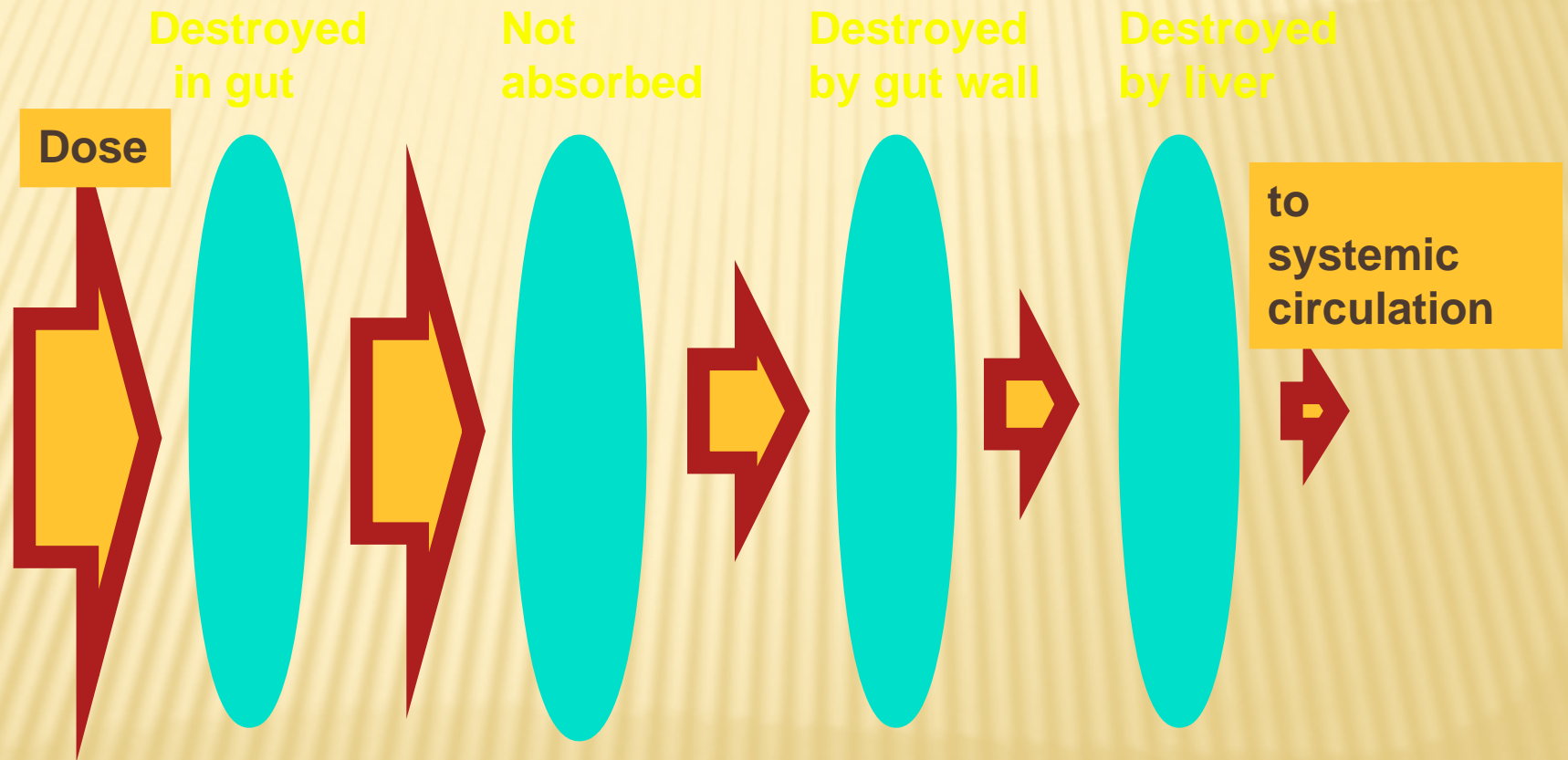
- ✗ the proportion of the drug in a dosage form available to the body

i.v injection gives 100% bioavailability.

Calculated from comparison of the area under the curve (AUC) relating plasma concentration to time for iv dosage compared with other route.

Says nothing about effectiveness.

BIOAVAILABILITY



SUSTAINED RELEASE PREPARATIONS

- ✗ depot injections (oily, viscous, particle size)
- ✗ multilayer tablets (enteric coated)
- ✗ sustained release capsules (resins)
- ✗ infusors (with or without sensors)
- ✗ skin patches (nicotine, GTN)
- ✗ pro-drugs
- ✗ liposomes

OVERVIEW - DISTRIBUTION

The body is a container in which a drug is distributed by blood (different flow to different organs) - but the body is not homogeneous. Note local delivery (asthma).

- Volume of distribution = $V = D/C_0$

plasma (3.5 l); extracellular fluid (14 l); intracellular fluid (50 l); + special areas (foetus, brain)

- note:::

plasma protein binding

tissue sequestration

---- brings drug to target tissue and affects concentration at site of action/elimination----

DISTRIBUTION INTO BODY COMPARTMENTS

- Plasma **3.5 litres**, heparin, plasma expanders
- Extracellular fluid **14 litres**, tubocurarine, charged polar compounds
- Total body water **40 litres**, ethanol
- Transcellular **small**, CSF, eye, foetus
(must pass tight junctions)

Plasma protein binding; Tissue

OVERVIEW - METABOLISM

- ✗ Drug molecules are processed by enzymes evolved to cope with natural compounds
- ✗ Drug may have actions increased or decreased or changed
- ✗ Individual variation genetically determined
- ✗ May be several routes of metabolism
- ✗ May not be what terminates drug action
- ✗ May take place anywhere BUT liver is prime site
- ✗ Not constant - can be changed by other drugs; basis of many drug-drug interactions

... metabolism is what the body does to the drug

BIOTRANSFORMATION OF DRUGS

- ✗ Mutations allowing de-toxification of natural toxic materials are advantageous and are selected
- ✗ Drugs are caught up in these established de-toxification processes
- ✗ Drugs may be converted to
 - less toxic/effective materials
 - more toxic/effective materials
 - materials with different type of effect or toxicity

SITES OF BIOTRANSFORMATION

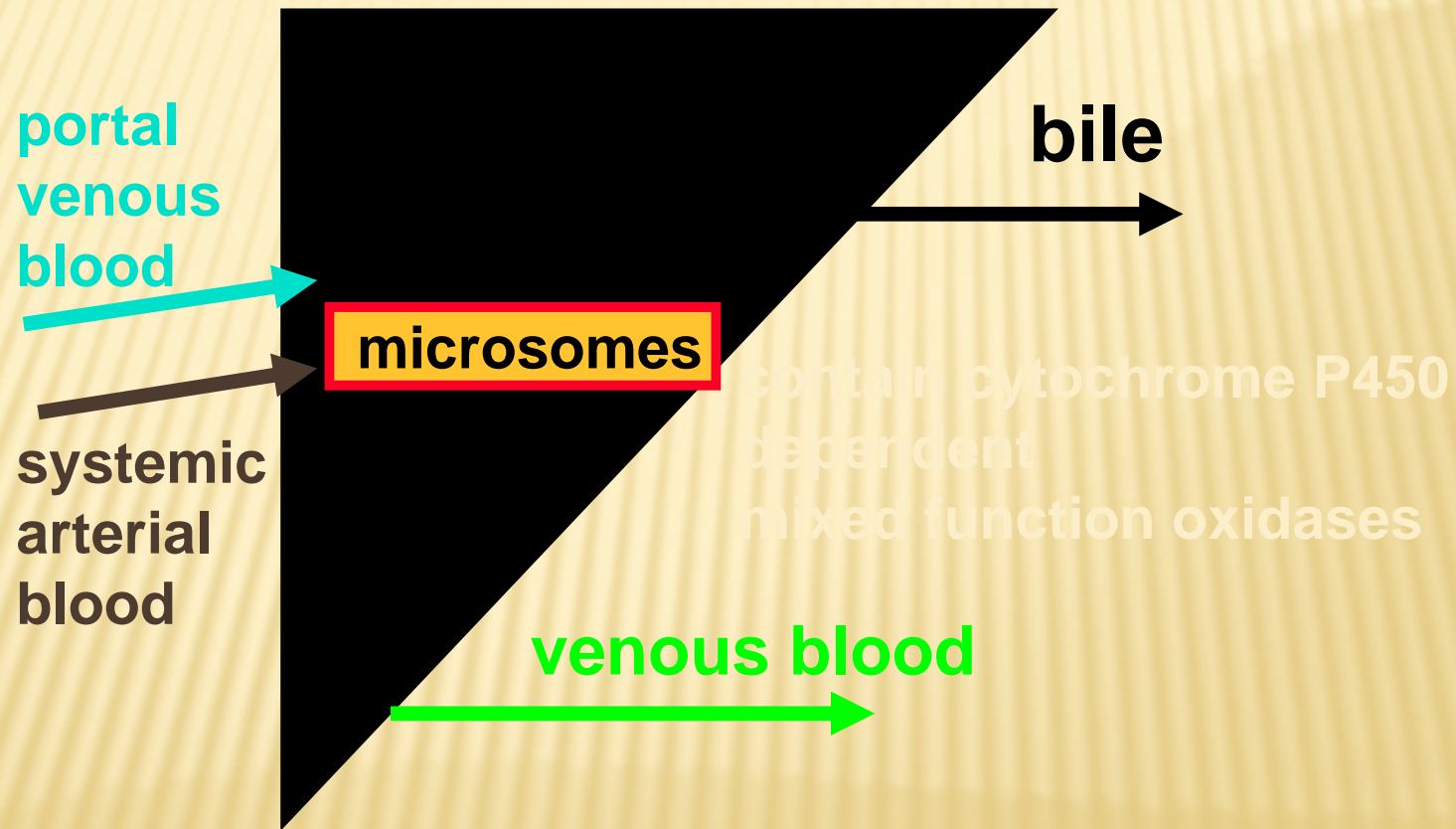
- ✗ where ever appropriate enzymes occur; plasma, kidney, lung, gut wall and



LIVER

- ✗ the liver is ideally placed to intercept natural ingested toxins (bypassed by injections etc) and has a major role in

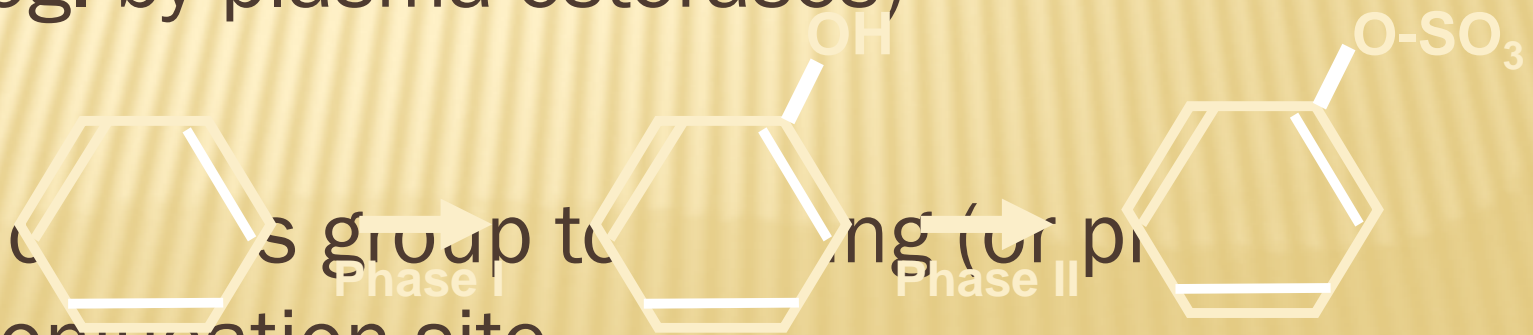
THE LIVER



TYPES OF BIOTRANSFORMATION REACTION

- ✖ Any structural change in a drug molecule may change its activity
- ✖ Phase I - changes drugs and creates site for phase II oxidation (adds O) eg. Microsomes (P450); reduction; hydrolysis (eg. by plasma esterases) others

- ✖ Phase II - conjugation (or phase I group to be formed) conjugation site



GENETIC POLYMORPHISM IN CYTOCHROME P450 DEPENDENT MIXED FUNCTION OXIDASES

CYP

FOUR families 1-4

SIX sub-families A-F

up to TWENTY isoenzymes 1-20

CYP3A4 : CYP2D6 : CYP2C9 : CYP2C19 :CYP2A6

**CYP2D6*17 (Thr107Ile; Arg296Cys) Caucasian 0%
Africans 6% Asian 51% - reduced affinity for
substrates**

PHASE 1 reactions

Hydroxylation $-\text{CH}_2\text{CH}_3 \rightarrow -\text{CH}_2\text{CH}_2\text{OH}$

Oxidation $-\text{CH}_2\text{OH} \rightarrow -\text{CHO} \rightarrow -\text{COOH}$

N-de-alkylation $-\text{N}(\text{CH}_3)_2 \rightarrow -\text{NHCH}_3 + \text{CH}_3\text{OH}$

Oxidative deamination $-\text{CH}_2\underset{\substack{| \\ \text{NH}_2}}{\text{CH}}\text{CH}_3 \rightarrow -\text{CHCOCH}_3 + \text{NH}_3$

PHASE 2 reactions

Conjugations with glucuronide, sulphate

.... alters activity, made less lipid soluble so excreted

PHASE 2 REACTIONS

(NOT ALL IN LIVER)

CONJUGATIONS

- ✗ -OH, -SH, -COOH, -CONH₂ give glucuronides
- ✗ -OH with sulphate to give sulphates
- ✗ -NH₂, -CONH₂, aminoacids, sulpha drugs with acetyl- to give acetylated derivatives
- ✗ -halo, -nitrate, epoxide, sulphate with glutathione to give glutathione conjugates

all tend to be less lipid soluble and therefore better excreted (less well reabsorbed)

OTHER (NON-MICROSOMAL) REACTIONS

- ✗ Hydrolysis in plasma by esterases (suxamethonium by cholinesterase)
- ✗ Alcohol and aldehyde dehydrogenase in cytosolic fraction of liver (ethanol)
- ✗ Monoamine oxidase in mitochondria (tyramine, noradrenaline, dopamine, amines)
- ✗ Xanthine oxidase (6-mercaptopurine, uric acid production)
- ✗ enzymes for particular drugs (tyrosine hydroxylase, dopa-decarboxylase etc)

FACTORS AFFECTING BIOTRANSFORMATION

- ✗ race (CYP2C9; warfarin (bleeding) phenytoin (ataxia) Losartan (less cleared but less activated as well); also fast and slow isoniazid acetylators, fast = 95% Inuit, 50% Brits, 13% Finns, 13% Egyptians).
- ✗ age (reduced in aged patients & children)
- ✗ sex (women slower ethanol metabolizers)
- ✗ species (phenylbutazone 3h rabbit, 6h horse, 8h monkey, 18h mouse, 36h man); biotransformation route can change
- ✗ clinical or physiological condition
- ✗ other drug administration (induction (not CYP2D6) or inhibition)
- ✗ food (charcoal grill ++CYP1A)(grapefruit juice --CYP3A)
- ✗ first-pass (pre-systemic) metabolism

INHIBITORS AND INDUCERS OF MICROSOMAL ENZYMES

- ✗ INHIBITORS cimetidine

prolongs action of drugs or inhibits action of those biotransformed to active agents (pro-drugs)

- ✗ INDUCERS barbiturates, carbamazepine shorten action of drugs or increase effects of those biotransformed to active agents

- ✗ BLOCKERS acting on non-microsomal enzymes (MAOI, anticholinesterase drugs)

OVERVIEW - EXCRETION

- ✗ Urine is the main but NOT the only route.
- ✗ Glomerular filtration allows drugs $<25\text{K MW}$ to pass into urine; reduced by plasma protein binding; only a portion of plasma is filtered.
- ✗ Tubular secretion active carrier process for cations and for anions; inhibited by probenacid.
- ✗ Passive re-absorption of lipid soluble drugs back into the body across the tubule cells.

Note effect of pH to make more of weak acid drug present in ionised form in alkaline pH therefore re-absorbed less and excreted faster; vica-versa for weak bases.

SPECIAL ASPECTS OF EXCRETION

- ✗ lactating women in milk
- ✗ little excreted in faeces unless poor formulation or diarrhoea
- ✗ volatile agents (general anaesthetics) via lungs
- ✗ the entero-hepatic shunt glucuronic acid conjugates with MW >300 are increasingly excreted in bile; hydrolysis of say -OH conjugate by beta-glucuronidase in gut will restore active drug which will be reabsorbed and produce an additional effect.

PHARMACOKINETICS

- ✗ Study of ADME on a quantitative basis

In man study blood, urine, faeces, expired air.

Measure urine volume & concentration of drug

$$\frac{\text{conc in urine} \times \text{vol per min}}{\text{plasma concentration}} = \text{RE} / \text{CL}$$

If neither secreted nor reabsorbed then clearance =
clearance of inulin = 120 ml/min

If completely cleared by secretion then clearance = clearance
of p-hippuric acid = renal blood flow = 700 ml/min

PHARMACOKINETIC PARAMETERS

- Volume of distribution $V = \text{DOSE} / C_0$
- Plasma clearance $Cl = K_{el} \cdot V$
- plasma half-life ($t_{1/2}$) directly from graph
or $t_{1/2} = 0.693 / K_{el}$
- Bioavailability $(AUC)_x / (AUC)_{iv}$

MULTIPLE DOSING

- In a dental context some drugs are given as single doses. Many however are given as a course of therapy
- On multiple dosing plasma concentration will rise and fall with each dose and will increase until
administration = elimination
ie. steady state is reached.
- At each dose the level will oscillate through a range
- The objective is to remain within the therapeutic window, with acceptable variation at each dose and with a regimen which promotes compliance.